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A novel total synthesis of the bioactive poly-substituted carbazole alkaloid carbazomadurin A

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1. Introduction

Naturally occurring carbazoles are very attractive compounds because of their antioxidant activity.¹ The neuronal cell-protecting carbazole alkaloids carbazomadurins A (1) and B (2) were isolated in 1997 from the microorganism Actinomadura madurae 2808-SV1 by Seto and co-workers (Fig. 1).² The structures of both compounds except the absolute configuration for 2 were assigned based on their spectroscopic data. These compounds protect against glutamate toxicity in neuronal hybridoma N18-RE-105 cells as an in vitro ischemia model. The Knölker group achieved the first total syntheses of carbazomadurin A $(1)^3$ and B $(2)^4$ by constructing a suitable carbazole framework based on a palladium-catalyzed sequence of the Buchwald-Hartwig amination, followed by oxidative cyclization and subsequent introduction of the alkenyl group at the C1 position using the Stille coupling reaction. In their first total synthesis, the Knölker group has also assigned the absolute configuration for 2, which was shown to have an S-configuration at the stereogenic center.4

We are interested in the synthesis of bioactive condensed heterocyclic compounds including natural products, based on a thermal electrocyclic reaction strategy with either a 6π -electron system or an aza 6π -electron system incorporating an aromatic or heteroaromatic double bond.^{5–7} In the course of our studies focusing on the development of a thermal electrocyclic reaction of a 6π -electron system including an allene intermediate,^{5,7} we

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ABSTRACT

A new total synthesis of the neuronal cell-protecting carbazole alkaloid carbazomadurin A is described. The key step was an allene-mediated electrocyclic reaction involving an indole [*b*]-bond for the construction of a highly substituted carbazole ring. The *E*-alkenyl side chain at the C1 position of carbazole was introduced between *O*-triflate and alkenyl pinacol borate using the Suzuki–Miyaura reaction. SEM groups were cleaved with TBAF and the formyl group was reduced to provide carbazomadurin A.

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achieved a new total synthesis of a highly functionalized carbazole alkaloid, carbazomadurin A (1). Our synthetic strategy of carbazomadurin A (1) is outlined in Scheme 1, where a 1,3,8-trioxygenated carbazole framework of 1 is obtained from 2-allenylindole intermediate **5** generated from 2-propargylindole **6**. We also hypothesized that an alkenyl side chain with an *E* configuration at the C1 position of 1 might be introduced by the Suzuki–Miyaura reaction⁸ using alkenyl pinacol borate **4**.

2. Results and discussion

We initially attempted a synthesis of the required 7-oxygenated 2,3,4,7-tetrasubstituted indole **13** from the known ethyl 7-isopropoxyindole-2-carboxylate (**7**)⁹ as shown in Scheme 2. Treatment of **7** with α , α -dichloromethyl methyl ether in the presence of TiCl₄ at -40 °C¹⁰ for 4 h gave 4-formylindole **8**. The formyl group of **8**



Figure 1.





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Scheme 1.



Scheme 2. Reagents and conditions: (a) $CH_3OCHCl_2 1 M TiCl_4$, CH_2Cl_2 , -40 °C, 4 h, 86%; (b) NaBH₄, EtOH, 0 °C, 5 h, 99%; (c) MOMCl, *i*Pr₂NEt, CH_2Cl_2 , 0 °C to rt, 48 h, 86%; (d) 65% Red-Al, toluene, 0 °C, 3 h, 99%; (e) MnO₂, CH_2Cl_2 , rt, 12 h, 84%; (f) KOH, I_2 , DMF, 0 °C, 12 h, 94%.

was subsequently reduced to a hydroxymethyl group with NaBH₄, which was treated with MOMCl in the presence of *N*,*N*-diisopropylethylamine (*i*Pr₂NEt) to yield MOM-ether **10**. Reduction of the ester moiety of **10** by Red-Al, followed by oxidation of the resulting alcohol by MnO₂ afforded the indole-2-carboxaldehyde **12**. Further treatment of **12** with I₂ in the presence of KOH gave the 7-oxygenated 2,3,4,7-tetrasubstituted indole **13** in six-steps.



Scheme 3. Reagents and conditions: (a) (1) Me₃Al, Cp₂ZrCl₂, 1,2-dichloroethane, rt, 12 h; (2) l_2 , THF, 0 °C to rt, 1 h, 78%; (b) bis(pinacolato)diboron, AcOK, PdCl₂(dppf), DMSO, 80 °C, 1 h, 48%.

On the other hand, a pinacol borate **4** to introduce an alkenyl side chain with an *E* configuration at the C1 of carbazomadurin A (**1**) was prepared from 5-methyl-1-hexyne (**14**) in three steps as follows (Scheme 3). Zirconium-catalyzed carboalumination of 5-methyl-1-hexyne (**14**) with trimethylaluminum in the presence



Scheme 4. Reagents and conditions: (a) ethoxyvinylstannane, $PdCl_2(PPh_3)_2$, Et_4NCl , DMF, 80 °C, 2 h, 86%; (b) ethynylmagnesium bromide, THF, 0 °C, 1 h, 81%; (c) MOMCl, iPr_2NEt , CH_2Cl_2 , 50 °C, 12 h, 81%; (d) TBAF, THF, 80 °C, 6 h, 40%; (e) DDQ, CH_2Cl_2 , rt, 20 min, 73%; (f) 4 M HCl, ethylene glycol, THF, 50 °C, 30 min, 91%; (g) PhNTf₂, NaH, THF, 0 °C, 30 min, 99%; (h) (1) BBr₃, CH_2Cl_2 , -78 °C to rt, 3 h; (2) SEMCl, iPr_2NEt , CH_2Cl_2 , 0 °C to rt, 12 h, 97% (two steps).

of zirconocene dichloride as reported by Negishi, ¹¹ followed by the addition of I₂ afforded the *E*-alkenyl iodide **15**³ in moderate yield. Subsequently, the Suzuki–Miyaura reaction^{8,12} of **15** with bis(pinacolato)diboron in the presence of PdCl₂(dppf) afforded the pinacol borate **4** in 48% yield.

We then synthesized 1,3,8-trioxygenated 1,2,3,5,8-pentasubstituted carbazoles 23 and 24 from the 2,3,4,7-tetrasubstituted indole 13 (Scheme 4). A Stille coupling reaction of 3-iodoindole 13 with 2ethoxyvinylstannane¹³ in the presence of PdCl₂(PPh₃)₂ gave the 3alkenylindole 16 in good yield. Grignard reaction of 16 with ethynylmagnesium bromide, followed by treatment of the resulting alcohol 17 with MOMCl and *i*Pr₂NEt produced the O-MOM-propargyl ether **18** in good yield. We subsequently attempted to construct a poly-functionalized carbazole framework 20 that was equivalent to **3**, using an allene-mediated electrocyclic reaction as a key step. When the O-MOM-propargyl ether **18** was heated at 90 °C in the presence of potassium tert-butoxide in tert-butyl alcohol according to the previously reported method⁷ for allene generation, the expected carbazole 20 was not detected. Alternatively, treatment of 18 with TBAF in THF according to another reported method^{7b} afforded the required carbazole 20 in somewhat low yield (40%). Oxidation of the O-MOM-methyl group¹⁴ at the C-5 of **20** with DDQ provided the 5-formylcarbazole 21, which was treated with 4 M HCl in the presence of ethylene glycol in THF to yield the 1hydroxycarbazole 22. Sequential treatment of 22 with N-phenylbis(trifluoromethanesulfonimide) (PhNTf₂) gave the corresponding



Scheme 5. Reagents and conditions: (a) Pd(PPh₃)₄, Na₂CO₃, DMF, 80 °C, 5 h, 99%; (b) Pd(PPh₃)₄, 3 M Na₂CO₃ aq., DMF, 70 °C, 2 h, 86%; (c) DIBAL-H, toluene, 0 °C, 1 h, 90%; (d) TBAF, THF, 80 °C, 1 h; (e) TBAF, HMPA, 100 °C, 1 h, 71%; (f) NaBH₄, MeOH, rt, 5 min, 70%.

triflate **23**. Cleavage of ethyl and isopropyl ethers of **23** with BBr₃ afforded the 3,8-dihydroxycarbazole, which was immediately protected with SEMCl and *i*Pr₂NEt to produce the 3,8-bis-O-SEM-carbazole **24**.

Carbon-carbon bond formations of two kinds of triflates 23 and 24 with the alkenyl pinacol borate 4 were then investigated (Scheme 5). The Suzuki-Miyaura cross-coupling reaction¹² of 23 with **4** in the presence of Na_2CO_3 and $Pd(PPh_3)_4$ gave the 1-alkenylcarbazole **25** in an excellent yield. In the case of cross-coupling 24 with 4, the coupling reaction proceeded smoothly in the presence of aqueous Na₂CO₃ and Pd(PPh₃)₄ to give **26** in good yield. Although, removal of the ethyl and isopropyl ethers of 25 with BBr₃ was attempted, the conversion to the corresponding **27** failed. By contrast, reduction of the formyl group of 26 with DIBAL gave the 5-hydroxymethylcarbazole **28** in good yield. Cleavage of both SEM groups of **28** using TBAF in THF did not provide carbazomadurin A (**1**). As a result, removal of both SEM groups¹⁵ of **26** with TBAF in HMPA gave the expected dihydroxycarbazole 27. Finally, reduction of **27** with NaBH₄ in methanol afforded carbazomadurin A (1). The physical and spectroscopic data¹⁶ of our synthetic carbazomadurin A (1) were identical with those of reported data.^{2,3}

3. Conclusions

The preparation of 2,3,4,7-tetrasubstituted indole 13 for the synthesis of poly-functionalized carbazoles 23 and 24 was achieved from 7 in a six-step sequence. An allene-mediated electrocyclic reaction of the propargyl ether **18** afforded the appropriate substituted carbazole 20, which led to the desired carbazoles 23 and 24. Introduction of the alkenyl side chain to the C1 position of carbazole O-triflates 23 and 24 was successfully achieved by the Suzuki-Mivaura cross-coupling reaction using alkenvl pinacol borate 4. Carbazomadurin A (1) was obtained from the 3,8-O-bisSEM carbazole 26 in two steps. Thus a new total synthesis of carbazomadurin A (1) was completed in two key steps by a synthesis of the suitable substituted carbazole, based on an allene-mediated electrocyclic reaction of a 6π -electron system involving an indole [b]bond and then the Suzuki-Miyaura reaction. This synthetic strategy is applicable for a series of other functionalized carbazole alkaloids.

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References and notes

- For most recent reviews, see: (a) Chakraborty, D. P.. In *The Alkaloids*; Cordell, G. A., Ed.; Academic Press: Amsterdam, 1993; Vol. 44, pp 257–364; (b) Knölker, H.-J.; Reddy, K. R. *Chem. Rev.* 2002, 102, 4303–4428; (c) Knölker, H.-J. *Top. Curr. Chem.* 2005, 244, 115–148; (d) Knölker, H.-J.; Reddy, K. R. In *The Alkaloids*; Cordell, G. A., Ed.; Academic Press: Amsterdam, 2008; Vol. 65, pp 1–430; (e) Knölker, H.-J. *Chem. Lett.* 2009, 38, 8–13.
- Kotoda, N.; Shinya, K.; Furihata, K.; Hayakawa, Y.; Seto, H. J. Antibiot. 1997, 50, 770-772.
- 3. Knölker, H.-J.; Knöll, J. Chem. Commun. 2003, 1170-1171.
- 4. Knöll, J.; Knölker, H.-J. Synlett 2006, 651-653.
- (a) Hibino, S.; Sugino, E.. In Advances in Nitrogen Heterocycles; Moody, C. J., Ed.; JAI Press: Greenwich, CT, 1995; Vol. 5, pp 205–227; (b) Kawasaki, r.; Sakamoto, M. J. Indian Chem. Soc. 1994, 71, 443–457; (c) Choshi, T. Yakugaku Zasshi 2001, 121, 487–495; (d) Choshi, T.; Hibino, S. Heterocycles 2009, 77, 85–97.
- (a) Ohmura, K.; Choshi, T.; Watanabe, S.; Satoh, Y.; Nobuhiro, J.; Hibino, S. Chem. Pharm. Bull. 2008, 56, 237–238; (b) Kohno, K.; Azuma, S.; Choshi, T.; Nobuhiro, J.; Hibino, S. Tetrahedron Lett. 2009, 50, 590–592. and related references cited therein.
- (a) Choshi, T.; Sada, T.; Fujimoto, H.; Sugino, E.; Hibino, S. *Tetrahedron Lett.* 1996, 37, 2593–2596; (b) Choshi, T.; Fujimoto, H.; Sugino, E.; Hibino, S. *Heterocycles* 1996, 43, 1847–1854; (c) Choshi, T.; Sada, T.; Fujimoto, H.; Nagayama, C.; Sugino, E.; Hibino, S. J. Org. Chem. 1997, 62, 2535–2543; (d)

Hagiwara, H.; Choshi, T.; Fujimoto, H.; Sugino, E.; Hibino, S. Chem. Pharm. Bull.
1998, 46, 1948–1949; (e) Hagiwara, H.; Choshi, T.; Fujimoto, H.; Sugino, E.;
Hibino, S. Tetrahedron 2000, 56, 5807–5811; (f) Hagiwara, H.; Choshi, T.;
Hobuhiro, J.; Fujimoto, H.; Hibino, S. Chem. Pharm. Bull. 2001, 49, 881–886; (g)
Hirayama, M.; Choshi, T.; Kumemura, T.; Tohyama, S.; Nobuhiro, T.; Hibino, S.
Heterocycles 2004, 63, 1765–1770; (h) Tohyama, S.; Choshi, T.; Matsumoto, K.;
Yamabuki, A.; Ikegata, K.; Nobuhiro, J.; Hibino, S. Tetrahedron Lett. 2005, 46, 5263–5264; (i) Nobuhiro, J.; Hirayama, M.; Choshi, T.; Kamoshita, S.;
Maruyama, Y.; Sukenaga, Y.; Ishizu, T.; Fujioka, H.; Hibino, S. Heterocycles 2006, 70, 491–499; (j) Yamabuki, A.; Fujinawa, H.; Choshi, T.; Tohyama, S.;
Matsumoto, K.; Ohmura, K.; Nobuhiro, J.; Hibino, S. Tetrahedron Lett. 2006, 47, 5859–5861; (k) Tohyama, S.; Choshi, T.; Azuma, S.; Fujioka, H.; Hibino, S.

 (a) Miyaura, N.; Suzuki, A. Chem. Rev. 1995, 95, 2457–2483; (b) Miyaura, N. Top. Curr. Chem. 2002, 219, 11–58.

- Suzuki, H.; Gyoutoku, H.; Yokoo, H.; Shinba, M.; Sato, Y.; Yamada, H.; Murakami, Y. Synlett 2000, 1196–1198.
- Condie, G. C.; Channon, M. F.; Ivory Kumar, A. J.; Kumar, N.; Black, D. S. Tetrahedron 2005, 61, 4989–5004.
- (a) van Horn, D. E.; Negishi, E. J. Am. Chem. Soc. 1978, 100, 2252–2254; (b) Negishi, E.; van Horn, D. E.; King, A. O.; Okukado, N. Synthesis 1979, 501; (c)

Rand, C. L.; van Horn, D. E.; Moore, M. W.; Negishi, E. J. Org. Chem. **1981**, 46, 4097–4100; (d) van Horn, D. E.; Negishi, E.; Yoshida, T. J. Am. Chem. Soc. **1985**, 107, 6639–6647.

- (a) Oh-e, T.; Miyaura, N.; Suzuki, A. Synlett **1990**, 221–223; (b) Miyaura, N.; Ishiyama, T.; Hayashi, H.; Ishikawa, M.; Satoh, M.; Suzuki, A. J. Am. Chem. Soc. **1989**, 111, 314–321.
- 13. Kazankova, M. A.; Protsenko, N. P.; Lutsenko, I. F. Russ. J. Gen. Chem. 1968, 38, 106–108.
- 14. Wang, W.; Li, T.; Atturdo, G. J. Org. Chem. 1997, 62, 6598-6602.
- 15. Leboff, A.; Carbonnelle, A.-C.; Alazard, J.-P.; Thal, C.; Kende, A. S. *Tetrahedron Lett.* **1987**, *28*, 4163–4164.
- 16. Data of carbazomadurin A (1): mp 166.5–168.5 °C; IR (ATR) v: 3480, 3420, 1640, 1580, 1430, 1370 cm⁻¹. ¹H NMR (300 MHz, acetone- d_6) δ : 8.56 (1H, br s), 8.37 (1H, br s), 7.77 (1H, br s), 7.57 (1H, s), 6.90 (1H, d, *J* = 7.7 Hz), 6.71 (1H, d, *J* = 7.7 Hz), 6.41 (1H, s), 5.01 (2H, s), 3.90 (1H, br s), 2.33 (2H, t, *J* = 7.7 Hz), 2.26 (3H, s), 1.64–1.73 (1H, m), 1.50–1.57 (5H, m), 0.99 (6H, d, *J* = 6.6 Hz). ¹³C NMR (75 MHz, acetone- d_6) δ : 149.9, 142.9, 142.7, 133.5, 130.3, 128.1, 123.4, 122.4, 122.3, 121.3, 120.3, 118.8, 109.6, 106.8, 64.0, 38.0, 37.7, 29.0, 22.8 (2C), 17.9, 13.5, MS *m/z*: 353 (M⁺). HR-MS (EI) *m/z*: 353.1975 (M⁺) (calcd for C₂₂H₂₇NO₃: 353.1991).